## Indirect precursors to perfluorooctanesulfonate (PFOS): **Environment tier II assessment**

8 March 2019

CAS Registry Numbers: 31506-32-8, 1691-99-2, 24448-09-7, 40630-61-3, 376-14-7, 423-82-5, 127133-66-8, 594864-11-6, 68298-62-4, 68329-56-6, 68555-90-8, 68555-91-9, 68555-92-0, 68568-77-4, 68797-76-2, 68867-60-7, 68909-15-9, 70776-36-2, 92265-81-1, 68227-96-3, 68298-78-2, 68299-39-8, 68586-14-1, 68867-62-9, 70900-40-2, 68081-83-4, 94313-84-5, 2991-51-7, 161074-58-4 185630-90-4, 253682-96-1, 91081-99-1, 192662-29-6, 68649-26-3, 68608-14-0, 68608-13-9, 306973-47-7, 306975-56-4, 3820-83-5, 2250-98-8, 30381-98-7, 67969-69-1, 29117-08-6, 37338-48-0, 68958-61-2, 1652-63-7, 38006-74-5, 67939-88-2, 68298-11-3, 94133-90-1, 57589-85-2, 68891-96-3, 75260-69-4, 94133-91-2, 944578-05-6

- Preface
- Disclaimer
- **Grouping Rationale**
- Chemical Identity
- Physical and Chemical Properties
- Import, Manufacture and Use
- **Environmental Regulatory Status**
- **Environmental Exposure**
- **Environmental Effects**
- Categorisation of Environmental Hazard
- Risk Characterisation
- **Key Findings**
- Recommendations
- **Environmental Hazard Classification**
- References

#### **Preface**

This assessment was carried out by staff of the National Industrial Chemicals Notification and Assessment Scheme (NICNAS) using the Inventory Multi-tiered Assessment and Prioritisation (IMAP) framework.

The IMAP framework addresses the human health and environmental impacts of previously unassessed industrial chemicals listed on the Australian Inventory of Chemical Substances (the Inventory).

The framework was developed with significant input from stakeholders and provides a more rapid, flexible and transparent approach for the assessment of chemicals listed on the Inventory.

Stage One of the implementation of this framework, which lasted four years from 1 July 2012, examined 3000 chemicals meeting characteristics identified by stakeholders as needing priority assessment. This included chemicals for which NICNAS already held exposure information, chemicals identified as a concern or for which regulatory action had been taken overseas, and chemicals detected in international studies analysing chemicals present in babies' umbilical cord blood.

Stage Two of IMAP began in July 2016. We are continuing to assess chemicals on the Inventory, including chemicals identified as a concern for which action has been taken overseas and chemicals that can be rapidly identified and assessed by using Stage One information. We are also continuing to publish information for chemicals on the Inventory that pose a low risk to human health or the environment or both. This work provides efficiencies and enables us to identify higher risk chemicals requiring assessment.

The IMAP framework is a science and risk-based model designed to align the assessment effort with the human health and environmental impacts of chemicals. It has three tiers of assessment, with the assessment effort increasing with each tier. The Tier I assessment is a high throughput approach using tabulated electronic data. The Tier II assessment is an evaluation of risk on a substance-by-substance or chemical category-by-category basis. Tier III assessments are conducted to address specific concerns that could not be resolved during the Tier II assessment.

These assessments are carried out by staff employed by the Australian Government Department of Health and the Australian Government Department of the Environment and Energy. The human health and environment risk assessments are conducted and published separately, using information available at the time, and may be undertaken at different tiers.

This chemical or group of chemicals are being assessed at Tier II because the Tier I assessment indicated that it needed further investigation.

For more detail on this program please visit: www.nicnas.gov.au.

#### **Disclaimer**

NICNAS has made every effort to assure the quality of information available in this report. However, before relying on it for a specific purpose, users should obtain advice relevant to their particular circumstances. This report has been prepared by NICNAS using a range of sources, including information from databases maintained by third parties, which include data supplied by industry. NICNAS has not verified and cannot guarantee the correctness of all information obtained from those databases. Reproduction or further distribution of this information may be subject to copyright protection. Use of this information without obtaining the permission from the owner(s) of the respective information might violate the rights of the owner. NICNAS does not take any responsibility whatsoever for any copyright or other infringements that may be caused by using this information.

Acronyms & Abbreviations

## **Grouping Rationale**

This Tier II assessment considers the environmental risks associated with the industrial uses of a group of 55 chemicals. The chemicals in this group are structurally related compounds in that they all contain a chain of eight perfluorinated carbons linked to a terminal sulfonamide or sulfonic ester group.

NICNAS has developed an action plan to assess and manage chemicals which may degrade to perfluorinated carboxylic acids, perfluoroalkyl sulfonates and similar chemicals (NICNAS, 2018a). The primary assumption outlined in this action plan is that chemicals with a perfluorinated chain terminated by a sulfonyl group will degrade to the perfluoroalkyl sulfonate. On this basis the chemicals in this group are each considered to have the potential to degrade into the environmentally hazardous perfluorooctanesulfonate anion (PFOS).

Perfluorooctanesulfonic acid, perfluorooctanesulfonates, perfluorooctanesulfonamides and perfluorooctanesulfonyls are listed on Annex III of the Rotterdam Convention on the Prior Informed Consent Procedure for Certain Hazardous Chemicals and

Pesticides in International Trade (the Rotterdam Convention). In addition, perfluorooctanesulfonic acid and certain PFOS-related substances have recently been identified as Persistent Organic Pollutants (POPs) under Annex B of the Stockholm Convention on Persistent Organic Pollutants (the Stockholm Convention). Further information on PFOS can be found in the Environment Tier II Assessment of the Direct Precursors to Perfluorooctanesulfonate (PFOS) group (NICNAS, 2015a).

The degradation of PFOS is very slow compared with its rate of formation from degradation of the precursors and PFOS will be the final degradant from multiple precursors. Therefore, the amount of PFOS in the environment (general or local) is expected to be higher than that of any of the precursors. It will therefore be assumed for the purposes of this assessment that the primary risk posed by the chemicals in this group results from the release of PFOS to the environment. The IMAP-Environment Tier II assessment for Direct Precursors to Perfluorooctanesulfonate (PFOS) (NICNAS, 2015a) has been used as a reference assessment.

#### **Chemical Identity**

In this assessment, "PFOS" is used to denote the conjugate base anion of perfluorooctanesulfonic acid (i.e. the perfluorooctanesulfonate anion). However, it is noted that this descriptor is commonly used in relation to a range of substances which may easily form the anion in water, such as the parent acid and salts of the acid.

Acronyms for certain chemicals in this group have been taken from a list of recommended acronyms and other terminology recently compiled to facilitate clear and unambiguous communication regarding the major classes of industrially important perfluoroalkyl and polyfluoroalkyl substances (Buck, et al., 2011).

This assessment includes a range of neutral and ionic organic chemicals, polymers, telomers, and chemicals that are in the category of unknown or variable composition, complex reaction products or biological materials (UVCBs). The chemicals in this assessment have been sub-grouped based on similar structural properties below for presentation purposes.

Structural information is presented where it is available. For polymers and telomers, constituent monomers are presented. Due to the typically variable size and structure of polymer molecules, the molecular formula and molecular weight for these substances are not presented.

Where insufficient structural information is available, the CAS RN and chemical name are presented only.

#### **Neutral Organic Derivatives of Perfluorooctanesulfonamides**

CAS RN	31506-32-8
Chemical Name	1-Octanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -methyl-
Synonyms	MeFOSA  N-methylperfluorooctanesulfonamide
Structural Formula	

	F HN CH <sub>3</sub>
Molecular Formula	C <sub>9</sub> H <sub>4</sub> F <sub>17</sub> NO <sub>2</sub> S
Molecular Weight (g/mol)	513.17
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)NC$
CAS RN	1691-99-2
Chemical Name	1-Octanesulfonamide, <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -(2-hydroxyethyl)-
Synonyms	EtFOSE  N-ethylperfluorooctanesulfonamidoethanol
Structural Formula	HO $N - S = O$ $F + F + F$ $F + F + F$ $F + F + F$
Molecular Formula	C <sub>12</sub> H <sub>10</sub> F <sub>17</sub> NO <sub>3</sub> S

Molecular Weight (g/mol)	571.25
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)N(CCO)CC$
CAS RN	24448-09-7
Chemical Name	1-Octanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -(2-hydroxyethyl)- <i>N</i> -methyl-
Synonyms	MeFOSE  N-methylperfluorooctanesulfonamidoethanol
Structural Formula	HO N - $\frac{1}{s} = 0$ F F F F F F F F F F F F F F F F F F F
Molecular Formula	C <sub>11</sub> H <sub>8</sub> F <sub>17</sub> NO <sub>3</sub> S
Molecular Weight (g/mol)	557.22
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)N(C)CCO$
CAS RN	40630-61-3

3/06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
Chemical Name	1-Octanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> , <i>N</i> -bis(2-hydroxyethyl)-
Synonyms	N,N-bis(2-hydroxyethyl)perfluorooctanesulfonamide
Structural Formula	HO N-S=O FFFFFFFFFFFFFFFFFFFFFFFFFFFFFFFFFFFF
Molecular Formula	C <sub>12</sub> H <sub>10</sub> F <sub>17</sub> NO <sub>4</sub> S
Molecular Weight (g/mol)	587.25
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)N(CCO)CCO$
CAS RN	376-14-7
Chemical Name	2-Propenoic acid, 2-methyl-, 2- [ethyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl ester
Synonyms	EtFOSMAC  N-ethylperfluorooctanesulfonamidoethyl methacrylate
Structural Formula	

00/2020	H <sub>2</sub> C = CH <sub>3</sub>
Molecular Formula	C <sub>16</sub> H <sub>14</sub> F <sub>17</sub> NO <sub>4</sub> S
Molecular Weight (g/mol)	639.32
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)S(=O)(=O)N(CCOC(=O)C(=C)C)CC$
CAS RN	423-82-5
Chemical Name	Acrylic acid, ester with <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -(2-hydroxyethyl)-1-octanesulfonamide
Synonyms	EtFOSAC  N-ethylperfluorooctanesulfonamidoethyl acrylate
Structural Formula	$H_{3}C$ $= 0$ $H_{3}C$ $F$ $F$ $F$ $F$ $F$ $F$
Molecular Formula	C <sub>15</sub> H <sub>12</sub> F <sub>17</sub> NO <sub>4</sub> S
Molecular Weight	625.30

28/0	6/2020 (g/mol)	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
	SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)S(=O)(=O)N(CCOC(=O)C=C)CC$

# Methacrylate and Acrylate Polymers with Perfluorooctanesulfonamide Side Chains

CAS RN	127133-66-8
Chemical Name	2-Propenoic acid, 2-methyl-, polymers with butyl methacrylate, lauryl methacrylate and 2-[methyl[(perfluoro-C4-8-alkyl)sulfonyl]amino]ethyl methacrylate
CAS RN	594864-11-6
Chemical Name	2-Propenoic acid, butyl ester, polymer with 2- [butyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl 2-propenoate and 2- methylpropyl 2-propenoate
CAS RN	68298-62-4
Chemical Name	2-Propenoic acid, 2-[butyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl ester, telomer with 2-[butyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, methyloxirane polymer with oxirane di-2-propenoate, methyloxirane polymer with oxirane mono-2-propenoate and 1-octanethiol
Synonyms	N-butylperfluorooctanesulfonamidoethyl acrylate telomer with N-butylperfluorooctanesulfonamidoethyl acrylate propylene glycol polymer with ethylene glycol-propylene glycol copolymer diacrylate with ethylene glycol-propylene glycol-propylene glycol copolymer monoacrylate and n-octyl mercaptan
Structural Formula	ation/imap-assessments/imap-assessments/tier-ii-environment-assessments/indirect-precursors-to-pe

/06/2020	Indirect precursors to pernuorooctanesulronate (PFOS): Environment tier ii assessment  H <sub>BC</sub>
SMILES	C(C=C)(=O)OCC[N@@](S(C(F)(F)C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(
CAS RN	68329-56-6
Chemical Name	2-Propenoic acid, eicosyl ester, polymer with 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, hexadecyl 2-propenoate, 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate and octadecyl 2-propenoate
CAS RN	68555-90-8
Chemical Name	2-Propenoic acid, butyl ester, polymer with 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, 2- [methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyll(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2-

[[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate and 2-[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl2-propenoate

Synonyms

butyl acrylate, polymer with N-methylperfluoroalkyl-C4-8-sulfonamidoethyl

06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment acrylates
Structural Formula	
SMILES	C(C=C)(=O)OCC[N@@](S(=O)(=O)C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(

CAS RN	68555-91-9
Chemical Name	2-Propenoic acid, 2-methyl-, 2- [ethyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl ester, polymer with 2- [ethyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate, 2- [ethyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate, 2- [ethyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate, 2- [ethyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate and octadecyl 2-methyl-2-propenoate
Synonyms	N-ethylperfluorooctanesulfonamidoethyl methacrylate, polymer with N-ethylperfluorooalkyl-C4-7-sulfonamidoethyl methacrylates and stearyl methacrylate
Structural Formula	

Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
HGC N CH <sub>3</sub>
P P P P P P P P P P P P P P P P P P P
H <sub>B</sub> C CH <sub>B</sub>
C(C(=C)C)(=O)OCCCCCCCCCCCCCCCCCCCCCCCCCC

CAS RN	68555-92-0
Chemical Name	2-Propenoic acid, 2-methyl-, 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl ester, polymer with 2- [methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate, 2- [methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate, 2- [methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate and octadecyl 2-methyl-2-propenoate
Synonyms	N-methylperfluorooctanesulfonamidoethyl methacrylate, polymer with N-methylperfluorooalkyl-C4-7-sulfonamidoethyl methacrylates and stearyl methacrylate
Structural Formula	

06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
SMILES	$CC(C(=O)OCC[N@@](C)S(=O)(=O)C(F)(F)C(F)(F)\\ C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)F=C.\\ CC(C(=O)OCC[N@@](C)S(=O)(=O)C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C($

CAS RN	68568-77-4
Chemical Name	2-Propenoic acid, 2-methyl-, 2- [ethyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl ester, polymer with 2- chloro-1,3-butadiene, 2-[ethyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2- methyl-2-propenoate, 2-[ethyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2- methyl-2-propenoate, 2-[ethyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2- methyl-2-propenoate and 2-[ethyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-methyl-2-propenoate

CAS RN	68797-76-2
Chemical Name	2-Propenoic acid, 2-methyl-, 2-ethylhexyl ester, polymer with 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, 2- [methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2-

[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-propenoate and oxiranylmethyl 2-methyl-2-propenoate

CAS RN	68867-60-7
Chemical Name	2-Propenoic acid, 2-[[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl ester, polymer with 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-propenoate and .alpha(1-oxo-2-propenyl)omegamethoxypoly(oxy-1,2-ethanediyl)
CAS RN	68909-15-9
Chemical Name	2-Propenoic acid, eicosyl ester, polymer with branched octyl 2-propenoate, 2-[[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-propenoate, octadecyl 2-propenoate and .alpha(1-oxo-2-propenyl)omegamethoxypoly(oxy-1,2-ethanediyl)
CAS RN	70776-36-2
Chemical Name	2-Propenoic acid, 2-methyl-, octadecyl ester, polymer with 1,1-dichloroethene, 2-[[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, N-(hydroxymethyl)-2-propenamide, 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate and 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate
Synonyms	stearyl methacrylate, polymer with 1,1-dichloroethene, n-methylolacrylamide and <i>N</i> -methylperfluoro-C4-8-sulfonamidoethyl acrylates

28/06/2020	Indirect precursors to perfluorooctanesulfona	ate (PFOS): Environment tier II assessmen
20/00/2020	indirect precursors to periluoroctaries unona	ate (i i oo). Environment tier ii assessinen

/06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
Structural Formula	
SMILES	C(C=C)(=O)OCC[N@@](S(=O)(=O)C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(C(
CAS RN	92265-81-1
Chemical Name	Ethanaminium, <i>N</i> , <i>N</i> , <i>N</i> -trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with 2-ethoxyethyl 2-propenoate, 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate and oxiranylmethyl 2-methyl-2-propenoate

# Methacrylate and Acrylate Telomers with Perfluorooctanesulfonamide Side Chains

CAS RN	68227-96-3
Chemical Name	2-Propenoic acid, butyl ester, telomer with 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, 2- [methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, .alpha(2-

methyl-1-oxo-2-propenyl)-.omega.-hydroxypoly(oxy-1,4-butanediyl), .alpha.-(2-methyl-1-oxo-2-propenyl)-.omega.-[(2-methyl-1-oxo-2propenyl)oxy]poly(oxy-1,4-butanediyl), 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate,2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-propenoate and 1octanethiol

CAS RN	68298-78-2
Chemical Name	2-Propenoic acid, 2-methyl-, 2-[[[5-[[2-[ethyl[(heptadecafluorooctyl)sulfonyl]amino] ethoxy]carbonyl]amino]-2-methylphenyl]amino] carbonyl]oxy]propyl ester, telomer with butyl 2-propenoate, 2-[[[5-[[2-[ethyl [(nonafluorobutyl)sulfonyl]amino]ethoxy]carbonyl] amino]-2-methylphenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2-[[[5-[[2-[ethyl [(pentadecafluoroheptyl)sulfonyl]amino]ethoxy] carbonyl]amino]-2-methylphenyl]amino]carbonyl]oxy] propyl 2-methyl-2-propenoate, 2-[[[5-[[[2-[ethyl [(tridecafluorohexyl)sulfonyl]amino]ethoxy]carbonyl] amino]-2-methylphenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2-[[[5-[[[2-[ethyl [(undecafluoropentyl)sulfonyl]amino]ethoxy]carbonyl] amino]-2-methylphenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2- [[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2-propenoate, 2- [methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(undecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2- [methyl[(undecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate and 1- octanethiol

CAS RN	68299-39-8
Chemical Name	2-Propenoic acid, 2-methyl-, 4- [[(heptadecafluorooctyl)sulfonyl]methylamino]butyl ester, telomer with butyl 2-propenoate, 2-[[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl 2- propenoate, 4-[methyl[(nonafluorobutyl)sulfonyl]amino]butyl 2-methyl-2- propenoate, 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, .alpha(2-methyl-1-oxo-2-propenyl)omegahydroxypoly(oxy-1,4- butanediyl), .alpha(2-methyl-1-oxo-2-propenyl)omega[(2-methyl-1-oxo-2-propenyl)oxy]poly(oxy-1,4-butanediyl), 4-

[methyl[(pentadecafluoroheptyl)sulfonyl]amino]butyl 2-methyl-2-propenoate, 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 4-[methyl[(tridecafluorohexyl)sulfonyl]amino]butyl 2-methyl-2-propenoate, 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 4-[methyl[(undecafluoropentyl)sulfonyl]amino]butyl 2-methyl-2-propenoate, 2-[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-propenoate and 1octanethiol

CAS RN	68586-14-1
Chemical Name	2-Propenoic acid, 2-[[(heptadecafluorooctyl)sulfonyl]methylamino]ethyl ester, telomer with 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, .alpha(2-methyl-1-oxo-2-propenyl)omegahydroxypoly(oxy-1,2-ethanediyl), .alpha(2-methyl-1-oxo-2-propenyl)omega[(2-methyl-1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl), 2-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate and 1-octanethiol
Synonyms	N-methylperfluorooctanesulfonamidoethyl acrylate, telomer with poly(ethylene glycol) dimethacrylate, poly(ethylene glycol) monomethacrylate, N-methylperfluoroalkyl-C4-7-sulfonamidoethyl acrylate and n-octyl mercaptan
Structural Formula	
SMILES	$ \begin{array}{l} O = C(OCCOCCOCC(=C)C)C(C) = C. \\ O = C(OCCOCCOCCO)C(C) = C.C(CCCCCCC)S. \\ C(C = C)(=O)OCC[N@@](S(C(F)(C(F)(F)C(F)(F))C(F)(F))C(F)(F)(F))C(F)(F)F)F)(F)F)(F)F)(F)F)(F)F)F)(F)F)F)(F)$

#### **Chemical Name**

carbonyl]amino]phenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2-[[[2-methyl-5-[[[4-[methyl[(pentadecafluoroheptyl)sulfonyl]amino]butoxy] carbonyl]amino]phenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2-[[[2-methyl-5-[[[4-[methyl[(tridecafluorohexyl)sulfonyl]amino]butoxy] carbonyl]amino]phenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2-[[[2-methyl-5-[[[4-[methyl[(undecafluoropentyl)sulfonyl]amino]butoxy] carbonyl]amino]phenyl]amino]carbonyl]oxy]propyl 2-methyl-2-propenoate, 2-[methyl[(nonafluorobutyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl](pentadecafluoroheptyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(tridecafluorohexyl)sulfonyl]amino]ethyl 2-propenoate, 2-[methyl[(undecafluoropentyl)sulfonyl]amino]ethyl 2-propenoate and 1octanethiol

#### **Carbamate Esters of Perfluorooctanesulfonamides**

CAS RN+	68081-83-4
Chemical Name	Carbamic acid, (4-methyl-1,3-phenylene)bis-, bis[2-[ethyl[(perfluoro-C4-8-alkyl)sulfonyl]amino]ethyl] ester
CAS RN	94313-84-5
Chemical Name	Carbamic acid, [5-[[[2-[[(heptadecafluorooctyl)sulfonyl]methylamino] ethoxy]carbonyl]amino]-2-methylphenyl]-, 9-octadecenyl ester, ( <i>Z</i> )-
Synonyms	(Z)-octadec-9-enyl [5-[[[2-[[(perfluorooctyl)sulphonyl]methylamino] ethoxy]carbonyl]amino]-o-tolyl]carbamate
Structural Formula	HG HG F F F F F F F F F F F F F F F F F
Molecular Formula	C <sub>38</sub> H <sub>50</sub> F <sub>17</sub> N <sub>3</sub> O <sub>6</sub> S
Molecular Weight (g/mol)	999.86
SMILES	$C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F)$ $C(F)(F)C(F)(F)F)S(=O)(=O)N(C)CCOC(=O)Nc1cc$ $(NC(=O)OCCCCCCC/C=C\setminus CCCCCCCC)c(C)cc1$

## **Carboxylic acid, Ester and Amide Derivatives of Perfluorooctanesulfonamides**

/06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
CAS RN	2991-51-7
Chemical Name	Glycine, <i>N</i> -ethyl- <i>N</i> -[(heptadecafluorooctyl)sulfonyl]-, potassium salt
Synonyms	potassium EtFOSAA  potassium <i>N</i> -ethylperfluorooctanesulfonamido acetate
Structural Formula	N-S=O H <sub>3</sub> C F F F F F F F F F F F F F F F F F F F
Molecular Formula	C <sub>12</sub> H <sub>7</sub> F <sub>17</sub> KNO <sub>4</sub> S
Molecular Weight (g/mol)	623.32
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)S(=O)(=O)N(CC(=O)[O-])CC.[K+]
CAS RN	75260-69-4
Chemical Name	Glycine, N-[(heptadecafluoroctyl)sulfonyl]-, monopotassium salt
Synonyms	potassium FOSAA

CAS RN	161074-58-4
Chemical Name	Fatty acids, C18-unsatd., trimers, reaction products with

28/06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment 1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8 heptadecafluoro- <i>N</i> -(2-hydroxyethyl)- <i>N</i> -methyl-1-octanesulfonamide, 1,1,2,2,3,3,4,4,4-nonafluoro- <i>N</i> -(2-hydroxyethyl)- <i>N</i> -methyl-1-butanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,7,7,7-pentadecafluoro- <i>N</i> -(2-hydroxyethyl)- <i>N</i> -methyl-1-heptanesulfonamide, 1,1,2,2,3,3,4,4,5,5,6,6,6-tridecafluoro- <i>N</i> -(2-hydroxyethyl)- <i>N</i> -methyl-1-hexanesulfonamide and 1,1,2,2,3,3,4,4,5,5,5-undecafluoro- <i>N</i> -(2-hydroxyethyl)- <i>N</i> -methyl-1-pentanesulfonamide
CAS RN	185630-90-4
Chemical Name	9-Octadecenoic acid ( <i>Z</i> )-, reaction products with <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -(2-hydroxyethyl)-1-octanesulfonamide
CAS RN	253682-96-1
Chemical Name	1-Octanesulfonamide, <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-, reaction products with succinic anhydride monopolyisobutylene derivs.
CAS RN	91081-99-1
Chemical Name	Sulfonamides, C4-8-alkane, perfluoro, <i>N</i> -(hydroxyethyl)- <i>N</i> -methyl, reaction products with epichlorohydrin, adipates (esters)
CAS RN	192662-29-6
Chemical Name	Sulfonamides, C4-8-alkane, perfluoro, <i>N</i> -[3-(dimethylamino)propyl], reaction products with acrylic acid

## **Urethane Polymers Incorporating Perfluorooctanesulfonamides**

CAS RN	68649-26-3

Chemical Name	Sulfonamides, C4-8 -alkane, perfluoro, <i>N</i> -(hydroxyethyl)- <i>N</i> -methyl, reaction products with 12-hydroxystearic acid and 2,4 -TDI, ammonium salts

28/06/2020

**Chemical Name** 

CAS RN

CAS RN

**CAS RN** 

**Chemical Name** 

**Chemical Name** 

CAS RN	306975-56-4
Chemical Name	Propanoic acid, 3-hydroxy- 2-(hydroxymethyl) -2-methyl-, polymer with 2-ethyl -2-(hydroxymethyl) -1,3-propanediol and <i>N</i> , <i>N</i> ', 2-tris(6-isocyanatohexyl) imidodicarbonic diamide reaction products with <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8 heptadecafluoro- <i>N</i> -(2-hydroxyethyl)-1-octanesulfonamide and <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,7-pentadecafluoro- <i>N</i> -(2-hydroxyethyl)-1-heptanesulfonamide, compds. with triethylamine

## **Phosphate Esters of Perfluorooctanesulfonamides**

CAS RN	3820-83-5
Chemical Name	1-Octanesulfonamide, <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -[2-(phosphonooxy)ethyl]-
Synonyms	N-ethyl perfluorooctanesulfonamidoethyl dihydrogen phosphate
Structural Formula	HO - P - O OH N - S = O F F F F F F F F F F F F F F F F F F
Molecular Formula	C <sub>12</sub> H <sub>11</sub> F <sub>17</sub> NO <sub>6</sub> PS
Molecular Weight (g/mol)	651.23
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)N(CCOP(=O)(O)O)CC$
CAS RN	2250-98-8
Chemical Name	1-Octanesulfonamide, <i>N,N',N''</i> -[phosphinylidynetris(oxy-2,1-ethanediyl)]tris[ <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-
Synonyms	tris(N-ethyl perfluorooctanesulfonamidoethyl)phosphate

Structural Formula	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
Molecular Formula	C <sub>36</sub> H <sub>27</sub> F <sub>51</sub> N <sub>3</sub> O <sub>10</sub> PS <sub>3</sub>
Molecular Weight (g/mol)	1757.70
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)S(=O)(=O)N(CCOP(=O) (OCCN(CC)S(=O)(=O)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)OCCN(CC) S(=O)(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)F)CC
CAS RN	30381-98-7
Chemical Name	1-Octanesulfonamide, <i>N</i> , <i>N</i> '-[phosphinicobis(oxy-2,1-ethanediyl)]bis[ <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro-, ammonium salt
Synonyms	ammonium bis( <i>N</i> -ethylperfluorooctanesulfonamidoethyl)phosphate
Structural Formula	
Molecular Formula	C <sub>24</sub> H <sub>22</sub> F <sub>34</sub> N <sub>3</sub> O <sub>8</sub> PS <sub>2</sub>
Molecular Weight (g/mol)	1221.50

SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)S(=O)(=O)N(CCOP(=O)([O-])OCCN(CC)S(=O)(=O)C(F) (F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)C(F)(F)C(F)(F)F)CC.[NH4+]

CAS RN	67969-69-1
Chemical Name	1-Octanesulfonamide, <i>N</i> -ethyl-1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-heptadecafluoro- <i>N</i> -[2-(phosphonooxy)ethyl]-, diammonium salt
Synonyms	diammonium N-ethylperfluorooctanesulfonamidoethyl phosphate
Structural Formula	
Molecular Formula	C <sub>12</sub> H <sub>17</sub> F <sub>17</sub> N <sub>3</sub> O <sub>6</sub> PS
Molecular Weight (g/mol)	685.29
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)F)S(=O)(=O)N(CCOP(=O)([O-])[O-])CC.[NH4+].[NH4+]

## Polyglycol Derivatives of Perfluorooctanesulfonamides

The substance represented by CAS RN 29117-08-6 is reported to contain 14 ethylene glycol units (Huawen and Vijayendran, 2001).

CAS RN	29117-08-6

28/06/2020	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
Chemical Name	Poly(oxy-1,2-ethanediyl), .alpha[2-
	[ethyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl]omegahydroxy-
Synonyms	N-ethylperfluorooctanesulfonamide poly(ethylene glycol)
Structural Formula	3) H.
Molecular Formula	C <sub>40</sub> H <sub>66</sub> F <sub>17</sub> NO <sub>17</sub> S
Molecular Weight (g/mol)	1187.99
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)F)S(=0)(=0)N(CCOCCOCCO CCOCCOCCOCCOCCOCCOCCOCCOCCOCCO
CAS RN	37338-48-0
Chemical Name	Poly[oxy(methyl-1,2-ethanediyl)], .alpha[2- [ethyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl]omegahydroxy-

The substance represented by CAS RN 68958-61-2 is a mixture of discrete chemicals. Representative chemical identity information is provided. However, it should be noted that this substance is expected to include a mixture of chemicals with polyethylene glycol chains containing between five and ten ethylene glycol units (average = 7.4) (Zou and Barton, 1994).

CAS RN	68958-61-2
Chemical Name	Poly(oxy-1,2-ethanediyl), .alpha[2- [ethyl[(heptadecafluorooctyl)sulfonyl]amino]ethyl]omegamethoxy-

28/0	6/2020 Synonyms	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment  N-ethylperfluorooctanesulfonamidoethyl poly(ethylene glycol) methyl ether
	Representative Structural Formula	He o do d
	Representative Molecular Formula	C <sub>27</sub> H <sub>40</sub> F <sub>17</sub> NO <sub>10</sub> S
	Representative Molecular Weight (g/mol)	893.65
	Representative SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)S(=O)(=O)N(CCOCCOCCOCCOCCOCCOCCOCCOCCOCCOCCOCCOCCO$

## Alkyl Ammonium and Amine Derivatives of Perfluorooctanesulfonamides

CAS RN	1652-63-7
Chemical Name	1-Propanaminium, 3-[[(heptadecafluorooctyl)sulfonyl]amino]- <i>N,N,N</i> -trimethyl-, iodide
Synonyms	N-(3-(trimethyl)aminopropyl)perfluorooctanesulfonamide iodide
Structural Formula	CH <sub>3</sub> H <sub>9</sub> C - N CH <sub>3</sub> N - S = 0 F F F F F F F F F F F F F F F F F F F

Molecular Formula	$C_{14}H_{16}F_{17}IN_2O_2S$
Molecular Weight (g/mol)	726.23
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)NCCC[N+](C)(C)C.[I-]$
CAS RN	38006-74-5
Chemical Name	1-Propanaminium, 3-[[(heptadecafluorooctyl)sulfonyl]amino]- <i>N,N,N</i> -trimethyl-, chloride
Synonyms	N-(3-(trimethyl)aminopropyl)perfluorooctanesulfonamide chloride
Structural Formula	$H_{0}C = N$ $CH_{3}$ $CH_{3}$ $CH_{3}$ $CH_{3}$ $CH_{3}$ $CH_{3}$ $CH_{3}$ $CH_{5}$ $CH_{7}$
Molecular Formula	$C_{14}H_{16}CIF_{17}N_2O_2S$
Molecular Weight (g/mol)	634.78
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) $C(F)(F)C(F)(F)F)S(=O)(=O)NCCC[N+](C)(C)C.[CI-]$
CAS RN	67939-88-2
1	

//06/2020 	Indirect precursors to perfluorooctanesulfonate (PFOS): Environment tier II assessment
	O-S=O  CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>5</sub> F  F  F  F  F  F  F  F  F  F  F  F  F
Molecular Formula	C <sub>18</sub> H <sub>23</sub> F <sub>17</sub> N <sub>2</sub> O <sub>6</sub> S <sub>2</sub>
Molecular Weight (g/mol)	750.49
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)S(=O)(=O)N(CCC[N+](C)(C) CCO)CCCS(=O)(=O)[O-]
CAS RN	94133-90-1
Chemical Name	1-Propanesulfonic acid, 3-[[3-(dimethylamino)propyl] [(heptadecafluorooctyl)sulfonyl] amino]-2-hydroxy-, monosodium salt
Synonyms	sodium <i>N</i> -(3-(dimethyl)aminopropyl) perfluorooctanesulfonamide 2-hydroxypropylsulfonate
Structural Formula	H <sub>0</sub> C N N S = 0 F F F F F F F F F F F F F F F F F F
Molecular Formula	$C_{16}H_{18}F_{17}N_2NaO_6S_2$

Molecular Weight (g/mol)	744.42
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)S(=O)(=O)N(CC(O)CS(=O)(=O)[O-])CCCN(C)C.[Na+]

CAS RN 94133-91-2 1-Propanaminium, 3-[[(heptadecafluorooctyl)sulfonyl](2-hydroxy-3-**Chemical Name** sulfopropyl)amino]-N-(2-hydroxyethyl)-N,N-dimethyl-, hydroxide, monosodium salt

## **Miscellaneous Perfluorooctanesulfonyl Derivatives**

CAS RN	57589-85-2
Chemical Name	Benzoic acid, 2,3,4,5-tetrachloro-6-[[[3- [[(heptadecafluorooctyl)sulfonyl]oxy]phenyl]amino]carbonyl]-, monopotassium salt
Synonyms	potassium 2,3,4,5-tetrachloro-6-[[[3- [[perfluorooctanesulfonyl]oxy]phenyl]amino]carbonyl] benzoate
Structural Formula	
Molecular Formula	$C_{22}H_5CI_4F_{17}KNO_6S$
Molecular Weight (g/mol)	915.22

SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)F)S(=O)(=O)Oc1cc(NC(=O)c2c(C(=O) [O-])c(Cl)c(Cl)c2Cl)ccc1.[K+]

The substance represented by CAS RN 68891-96-3 is a dinuclear coordination complex of chromium. The chromium ions in this complex are present in the 3+ oxidation state.

CAS RN	68891-96-3
Chemical Name	Chromium, diaquatetrachloro[.mu[ <i>N</i> -ethyl- <i>N</i> - [(heptadecafluorooctyl)sulfonyl]glycinato-O1:O1']]muhydroxybis[2- methylpropanol]di-
Synonyms	N-ethylperfluorooctanesulfonylamido acetate chromium complex
Structural Formula	
Molecular Formula	$C_{20}H_{32}CI_4Cr_2F_{17}NO_9S$
Molecular Weight (g/mol)	1031.31
SMILES	C(F)(F)(C(F)(F)C(F)(F)C(F)(F)C(F)(F) C(F)(F)C(F)(F)F)S(=O)(=O)N(CC1O[Cr+++] ([OH]CC(C)C)([OH2])([CI-])([CI-])[OH-][Cr+++]([OH2])([OH]CC(C)C)([CI-]) ([CI-])O=1)CC

CAS RN 944578-05-6 Sulfonamides, C4-8-alkane, perfluoro, N-[4,7-dimethyl-4-[[(1methylpropylidene)amino]oxy]-3,5-dioxa-6-aza-4-silanon-6-en-1-yl]-N-ethyl

#### **Physical and Chemical Properties**

Limited measured physical and chemical property data are available for the chemicals in this group.

The substances in this group include a wide range of neutral and ionic organic chemicals, polymers, telomers, UVCBs, and a coordination complex. Based on the available data, all are expected to be solids under ambient conditions (Hekster, et al., 2002; LMC, 2013). The higher molecular weight chemicals, including the telomers, polymers, and the chromium coordination complex are all expected to be involatile. However, the relatively lower molecular weight neutral organic substances may be volatile. Data available for EtFOSE indicate moderate volatility, with a measured vapour pressure of 0.504 Pa (Hekster, et al., 2002).

The measured water solubility value available for EtFOSE (0.151 mg/L) indicates that some chemicals in this group may be slightly soluble in water (Hekster, et al., 2002). When considered alongside volatility data, these values suggest some chemicals in this group may be moderately volatile from water (for example, estimated Henry's law constant for EtFOSE = 58 Pa-m<sup>3</sup>/mol) (US EPA, 2008).

#### Import, Manufacture and Use

#### Australia

No Australian manufacture or export of PFOS related chemicals is known (NICNAS, 2018b; Parliament of Australia, 2014). No import data specific to the individual chemicals in this group have been identified.

It is noted that the chemicals in this group may be present in the environment due to historic use or release from pre-treated articles imported into Australia. However, release from these uses is beyond the scope of this assessment.

#### International

Due to their unique chemical properties, perfluorinated chemicals have found use in a wide range of industrial applications. In the early 2000s, it was estimated that almost 500 tonnes of PFOS-based chemicals were used in the EU per annum, with significant uses including fabric treatments (49%), paper treatments (33%) and coating products, such as paints (18%) (Brooke, et al., 2004). However, use of PFOS containing chemicals has been declining globally. Concerns regarding persistence and bioaccumulation resulted in the largest global manufacturer of PFOS-based chemicals ceasing production of these chemicals in 2002 (3M, 2014; OECD, 2002). Nevertheless, production by various additional companies increased shortly after the phase out by the main manufacturer (US EPA, 2009).

Available data indicate that the chemicals in this group have multiple specialised applications. While the alcohols (e.g. EtFOSE and MeFOSE) and phosphate esters in this group have traditionally been used in fabric and paper treatments, more recent uses appear to predominantly be in paints, lacquers and varnishes (Hekster, et al., 2002; INERIS, 2013; Nordic Council of Ministers, 2015). In 2010, MeFOSE was reported to have been used in over 40 products in two Nordic countries (Nordic Council of Ministers, 2015).

The acetate salt (CAS RN 2991-51-7) and polyglycols (e.g. CAS RNs 29117-08-6 and 68958-61-2) have previously been used in industrial surfactant formulations (Flick, 1993). Use of the acetate has recently been reported in surface treatments, cleaning products and floor waxes (Nordic Council of Ministers, 2015). The alkyl ammonium derivatives are expected to have similar uses, with use of the quaternary ammonium salt (CAS RN 1652-63-7) reported in various Nordic countries in 2012 (Nordic Council of Ministers, 2015).

The reactive acrylate, EtFOSAC (CAS RN 423-82-5), and methacrylate, EtFOSMAC (CAS RN 376-14-7), are both expected to be used primarily as monomers in the manufacture of polymers and telomers (Buck, et al., 2011; Klun, et al., 2008; Smith, 1997). Use of a number of the resultant polymers and telomers has been reported recently. One acrylate polymer (CAS RN 68298-62-4) was reported to be present in over 70 preparations across two Nordic countries in 2012, with the majority being paint and coating products for motor vehicle bodywork. Other polymers and telomers are expected to have uses as impregnation agents (particularly in textiles), with one polymer (CAS RN 68555-92-0) identified as being used for this purpose in two Nordic countries in 2012 (KEMI, 2006; Nordic Council of Ministers, 2015).

#### **Environmental Regulatory Status**

#### **Australia**

In accordance with regulation 11C(1) under the Industrial Chemicals (Notification and Assessment) Regulations 1990 (Cwlth), introduction or export of chemicals listed on the Rotterdam Convention is prohibited unless approval is obtained from the NICNAS Director (Commonwealth of Australia, 1990).

In 2008, a factsheet published by NICNAS recommended that PFOS-based chemicals be restricted to essential uses only, and that importers ensure that alternative chemicals are less toxic and not persistent in the environment (NICNAS, 2018b).

#### **United Nations**

Perfluorooctanesulfonic acid, perfluorooctanesulfonates, perfluorooctanesulfonamides and perfluorooctanesulfonyls are listed on Annex III of the Rotterdam Convention (UNEP, 2013). The Rotterdam Convention aims to facilitate sharing of chemical information to promote shared responsibility for the international trade of certain hazardous chemicals. Of the 193 United Nations member countries, 153 have ratified the Rotterdam Convention, including Australia. The European Union is also a party to the convention (UNEP, 2014).

The chemicals in this group are not currently identified as Persistent Organic Pollutants (UNEP, 2001) or ozone depleting substances (UNEP, 1987).

#### **OECD**

The chemicals in this group have not been identified as High Production Volume chemicals, or sponsored for assessment under the Cooperative Chemicals Assessment Programme (OECD, 2013).

#### Canada

Chemicals containing the PFOS moiety, including neutral and ionic organic chemicals, polymers, telomers and UVCBs, are listed under Schedule 1 (the Toxic Substances List) of the Canadian Environmental Protection Act 1999. The import, manufacture, sale and use of these chemicals are prohibited, unless destined for an exempted application. Such applications include certain uses in photolithography, in aviation hydraulic fluid, as fume suppressants and in aqueous film forming foam at concentrations less than 0.5 parts per million (Government of Canada, 2008).

The majority of the chemicals in this group are not listed on the Canadian Domestic Substances List (DSL) (Government of Canada, 2013). The 13 chemicals which are listed were found to be Persistent (P) and Inherently Toxic to the Environment (iT<sub>E</sub>) during the Categorization of the DSL. Two polymers (CAS RNs 68329-56-6 and 70776-36-2) and the chromium complex (CAS RN 68891-96-3) were found to be Not Bioaccumulative (Not B), while the two quaternary ammonium surfactants (CAS RNs 1652-63-7 and 38006-74-5), two simple sulfonamide alcohols (CAS RNs 24448-09-7 and 1691-99-2) and an alcohol phosphate derivative (CAS RN 67969-69-1) were found to be Bioaccumulative (B). The bioaccumulation potentials of the remaining chemicals on the DSL were unable to be determined (Environment and Climate Change Canada, 2019).

#### **European Union**

The use of chemicals containing the PFOS moiety, including neutral and ionic organic chemicals, polymers, telomers and UVCBs, is restricted under Commission Regulation No 850/2004 on Persistent Organic Pollutants (European Commission, 2010). These chemicals may only be used in select applications in electroplating systems, photolithography processes, photographic coatings, chromium plating, and aviation hydraulic fluids. Use is to be phased out as alternate substances or technologies become available.

Twenty-seven chemicals in this group have been pre-registered for use in the European Union under the Registration, Evaluation, Authorisation and Restriction of Chemicals (REACH) legislation (ECHA, 2015). However, no chemicals in this group have yet undergone the full registration process (ECHA, 2014).

#### **United States of America**

Forty-five chemicals in this group, including neutral and ionic organic chemicals, polymers, telomers and UVCBs, are listed on the inventory of chemicals manufactured or processed in the USA, as published under the Toxic Substances Control Act 1976 (US EPA, 2014). All chemicals listed are subject to Significant New Use Rules. Approval is required before any of these chemicals may be used for any new applications as outlined under Section 721.9582 of the Code of Federal Regulations (US EPA, 2011).

The United States Environmental Protection Agency (US EPA) published an action plan on long-chain perfluorinated chemicals, covering the chemicals in this group, in 2009. All chemicals were identified as persistent, bioaccumulative and toxic (US EPA, 2009).

#### **Environmental Exposure**

Based on international use pattern information, chemicals in this group may have industrial uses in Australia. Release to the environment may occur during the use and disposal of formulations containing these chemicals.

The environmental degradant of primary concern for the chemicals in this group is PFOS. Biodegradation data available for EtFOSE demonstrate conversion to the corresponding carboxylic acid, N-ethylperfluorooctanesulfonamidoacetic acid (EtFOSAA), with the ultimate biodegradation product being PFOS (Hekster, et al., 2002; Martin, et al., 2010). This perfluorinated anion is purely of anthropogenic origin and has been identified as a POP under the Stockholm Convention (NICNAS, 2015a). It is widely assumed that analogous perfluorosulfonamides will be susceptible to similar biotransformation processes (Martin, et al., 2010).

Studies conducted with the shorter-chain perfluorobutanesulfonate homologue of MeFOSE also indicate potential for atmospheric degradation to PFOS. Oxidation by hydroxyl radicals was found to produce traces of the sulfonate, as well as aldehyde, sulfonamide and carboxylic acid derivatives. However, detailed information on the yield is not available (D'eon, et al., 2006; Martin, et al., 2010).

Monitoring data have identified some of the chemicals in this group and their degradation products in the environment. The two volatile alcohols (MeFOSE and EtFOSE) are routinely detected in outdoor air. Highest concentrations were found in air sampling conducted in Toronto, Canada in 2001, with the chemicals found in mean concentrations of 1.0 × 10<sup>-4</sup> micrograms per cubic metre ( $\mu$ g/m<sup>3</sup>) and 2.1 × 10<sup>-4</sup> $\mu$ g/m<sup>3</sup>, respectively (Martin, et al., 2010). In surface water, EtFOSAA has been measured at concentrations up to 7.5 nanograms per litre (ng/L) (Martin, et al., 2010).

Further, PFOS has been found to be a widely distributed environmental pollutant. International studies have identified PFOS in various environmental media, including surface and drinking waters. In Australia, PFOS has been found in drinking water at concentrations up to 16 ng/L (Thompson, et al., 2011). However, it is noted there are multiple potential sources of PFOS in the environment, including past industrial use of direct precursors to PFOS or use of PFOS-related substances in articles, in addition to the industrial use of derivatives of PFOS which degrade to this substance in the environment.

#### **Environmental Effects**

No ecotoxicity data were located for the parent chemicals in this group.

The currently available ecotoxicity data for PFOS are summarised in the IMAP-Environment Tier II assessment of Direct Precursors to PFOS. Data currently available for PFOS indicate moderate acute aguatic toxicity (median lethal/effective concentration values = 3.6-48.2 mg/L) and chronic toxicity (no-observed-effect concentration values = 0.25-7 mg/L) in standard ecotoxicity tests. However, data from a growing number of non-standard ecotoxicity tests have identified intergenerational toxicity in offspring when parent generations are exposed to concentrations as low as 0.01 mg/L (NICNAS, 2015a).

It is noted that the chemicals in this group contain a range of functional groups and moieties which are correlated with acute toxic effects in aquatic organisms. However, these possible effects have not been considered in this assessment, as the primary risk posed by the chemicals in this group is assumed to result from the cumulative release of PFOS into the environment.

#### **Categorisation of Environmental Hazard**

Insufficient data are presented in this assessment to categorise the chemicals in this group according to domestic environmental hazard thresholds (EPHC, 2009).

It is noted that direct precursors to PFOS are categorised as Persistent, Bioaccumulative and Toxic (PBT) substances according to domestic environmental hazard criteria (NICNAS, 2015a).

#### **Risk Characterisation**

The chemicals in this group may degrade to a PBT substance. It is not currently possible to derive a safe environmental exposure level for such chemicals and it is therefore not appropriate to characterise the environmental risks for these chemicals in terms of a risk quotient.

The recalcitrant degradant assumed to be formed from each of the chemicals in this group (PFOS) has been identified as a POP under the Stockholm Convention. Persistent Organic Pollutants are chemical substances that persist in the environment, bioaccumulate in humans and wildlife, undergo long-range transport to regions where they have never been used or produced, and pose a risk of causing harmful effects on human health and the environment. The international community has now, on several occasions, called for urgent global actions to reduce and/or eliminate production and use of such chemicals.

## **Key Findings**

Based on international data, chemicals in this group may have current industrial use in Australia in a range of products.

Available environmental degradation data show that some chemicals in this group degrade to PFOS. Therefore, the principal risk posed by the chemicals in this group if emitted to the environment has been assumed to result from the cumulative releases of PFOS. This perfluorinated chemical has been identified as a POP under the Stockholm Convention.

The chemicals in this group have not been categorised according to domestic environmental hazard criteria. This finding does not indicate a lack of potential hazard for some parent chemicals in this group, but rather a lack of suitable data to characterise these hazards for each individual chemical.

The assessment of the risks from these chemicals to human health, including public health, has been considered in the IMAP-Human Health Tier II assessment for Indirect Precursors to Perfluorooctanesulfonate (PFOS) (NICNAS, 2015b).

#### Recommendations

The chemicals in this group have been assessed as having the potential to give rise to adverse outcomes for the environment and public health. These chemicals are currently listed on the Australian Inventory of Chemical Substances (the Inventory), and are available to be introduced into Australia without any further assessment by NICNAS. Other chemicals with reduced potential for adverse outcomes are becoming available but, given the properties of these chemicals, their assessment as new chemicals

under the Industrial Chemicals (Notification and Assessment) Act 1989 (the ICNA Act) is still required to fully characterise the human health and the environmental risks associated with their use.

It is recommended that NICNAS consult with industry and other stakeholders to consider strategies, including regulatory mechanisms available under the ICNA Act, to encourage the use of safer chemistry.

#### **Environmental Hazard Classification**

Insufficient data are presented in this assessment to classify the aquatic hazards of chemicals in this group according to the third edition of the United Nations' Globally Harmonised System of Classification and Labelling of Chemicals (GHS) (UNECE, 2009).

It is noted that direct precursors to PFOS have been classified as Acute Aquatic Category 2 (H401) and Chronic Aquatic Category 1 (H410) under the GHS (NICNAS, 2015a).

#### References

3M (2014). 3M's Phase Out and New Technologies. 3M Company, St. Paul, USA. Accessed 12 November 2014 at http://solutions.3m.com.

Brooke D, Footitt A and Nwaogu TA (2004). Environmental Risk Evaluation Report: Perfluorooctanesulphonate (PFOS). Environment Agency, Wallingford, UK. Accessed 18 November 2014 at http://www.pops.int.

Buck RC, Franklin J, Berger U, Conder JM, Cousins IT, de Voogt P, Hensen AA, Kannan K, Mabury SA and van Leeuwen SP (2011). Perfluoroalkyl and Polyfluoroalkyl Substances in the Environment: Terminology, Classification and Origins. Integrated Environmental Assessment and Management, 7(4),pp 513-541.

Commonwealth of Australia (1990). Industrial Chemicals (Notification and Assessment) Regulations 1990.

D'eon JC, Hurley MD, Wallington TJ and Mabury SA (2006). Atmospheric Chemistry of N-methyl Perfluorobutane Sulfonamidoethanol, C<sub>4</sub>F<sub>9</sub>CO<sub>2</sub>N(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>2</sub>OH: Kinetics and Mechanism of Reaction with OH. Environmental Science and Technology, 40(6),pp 1862-1868.

ECHA (2014). Registered Substances. European Chemicals Agency, Helsinki, Finland. Accessed 9 May 2014 at http://echa.europa.eu.

ECHA (2015). Pre-registered substances. European Chemicals Agency, Helsinki, Finland. Accessed 1 April 2015 at http://echa.europa.eu.

Environment and Climate Change Canada (2019). Categorization Results from the Canadian Domestic Substance List. Provided by OECD. Accessed 4 January 2019 https://canadachemicals.oecd.org.

EPHC (2009). Environmental Risk Assessment Guidance Manual for Industrial Chemicals. Environment Protection and Heritage Council, Canberra, Australia. Accessed 9 December 2013 at http://www.scew.gov.au.

European Commission (2010). Commission Regulation (EU) No 757/2010 of 24 August 2010 amending Regulation (EC) No 850/2004 of the European Parliament and of the Council on persistent organic pollutants as regards Annexes I and III. Official Journal of the European Union, 223,pp 29-36.

Flick EW (1993). Industrial Surfactants: An Industrial Guide (2nd ed). Noyes Publications, New Jersey, USA. Accessed at https://books.google.com.au/books?id=oOSTN4gZnXYC.

Government of Canada (2008). Perfluorooctane Sulfonate and its Salts and Certain Other Compounds Regulations. Accessed 26 February 2015 at http://laws-lois.justice.gc.ca.

Government of Canada (2013). Search Engine for Chemicals and Polymers. Environment Canada, Gatineau, Canada. Accessed 1 May 2014 at http://www.ec.gc.ca.

Hekster F, de Voogt P, Pijnenburg A and Laane R (2002). Perfluoroalkylated substances: Aquatic environmental assessment, Report RIKZ/2002.043. Government of the Netherlands Ministry of Infrastructure and the Environment, Amsterdam, The Netherlands. Accessed 22 December 2014 at http://repository.tudelft.nl.

Huawen L and Vijayendran B (2001). Durable coating composition, process for producing durable, antireflective coatings, and coated articles. USA. Accessed 5 March 2015 at http://www.google.com.

INERIS (2013). Precursors of PFOS [French]. L'Institut National de l'Environnement Industriel et des Risques, Paris, France. Accessed 22 December 2014 at http://www.ineris.fr.

KEMI (2006), Perfluorinated substances and their uses in Sweden, Swedish Chemicals Agency (Swedish: Kemikalieinspektionen), Stockholm, Sweden. Accessed 6 March 2015 at http://www.kemi.se.

Klun TP, Kirk AR, Moore GGI, Clark JC and Qiu Z-M (2008). Process for Preparing Fluorochemical Monoisocyanates. USA. Accessed 5 January 2015 at https://drive.google.com/viewerng/viewer? url=patentimages.storage.googleapis.com/pdfs/US7361782.pdf.

LMC (2013). The OECD QSAR Toolbox for Grouping Chemicals into Categories, v 3.1. Laboratory of Mathematical Chemistry, University "Prof. Dr. Assen Zlatarov", Burgas, Bulgaria. Available at http://oasis-lmc.org.

Martin JW, Asher BJ, Beesoon S, Benskin JP and Ross MS (2010). PFOS or PreFOS? Are perfluorooctane sulfonate precursors (PreFOS) important determinants of human and environmental perfluorooctane sulfonate (PFOS) exposure? Journal of Environmental Monitoring, 12(11),pp 1979-2004.

NICNAS (2015a). Environment Tier II Assessment for Direct precursors to perfluorooctanesulfonate (PFOS). National Industrial Chemicals Notification and Assessment Scheme, Sydney, Australia. Accessed 21 April 2015 at http://www.nicnas.gov.au

NICNAS (2015b). IMAP Human Health Tier II Assessment for Indirect Precursors to Perfluorooctanesulfonate (PFOS). National Industrial Chemicals Notification and Assessment Scheme, Sydney, Australia. Accessed 21 April 2015 at http://www.nicnas.gov.au

NICNAS (2018a). Data requirements for notification of new chemicals containing a perfluorinated carbon chain. National Industrial Chemicals Notification and Assessment Scheme, Sydney, Australia. Accessed 3 January 2019 at http://www.nicnas.gov.au.

NICNAS (2018b), PFC Derivatives and Chemicals on which they are based: Alert Factsheet. National Industrial Chemicals Notification and Assessment Scheme, Sydney, Australia. Accessed 29 January 2019 at http://www.nicnas.gov.au.

Nordic Council of Ministers (2015). Substances in Preparations in Nordic Countries (SPIN). Chemical Group, Nordic Council of Ministers, Copenhagen, Denmark. Accessed 17 March 2015 at http://www.spin2000.net.

OECD (2002). Hazard Assessment of Perfluorooctane sulfonate (PFOS) and its Salts. Organisation for Economic Cooperation and Development, Paris, France. Accessed 11 November 2014 at http://www.oecd.org.

OECD (2013). OECD Existing Chemicals Database. Organisation for Economic Cooperation and Development, Paris, France. Accessed 13 November 2013 http://webnet.oecd.org.

Parliament of Australia (2014). Joint Standing Committee on Treaties: Explanatory Statement 2 of 2014: Amendments, Adopted on 10 May 2013, to Annex III of the Rotterdam Convention of the Prior Informed Constent Procedure for Certain Hazardous Chemicals and Pesticides in International Trade. Parliament of Australia, Canberra, Australia. Accessed 4 December 2014 at http://www.austlii.edu.au.

Smith RS (1997). Durable Repellent Fluorochemical Compositions. USA. Accessed 6 January 2015 at https://drive.google.com/viewerng/viewer?url=patentimages.storage.googleapis.com/pdfs/US5672651.pdf.

Thompson J, Eaglesham G and Mueller J (2011). Concentrations of PFOS, PFOA and other perfluorinated alkyl acids in Australian drinking water. Chemosphere, 83(10),pp 1320-1325.

UNECE (2009). Globally Harmonised System of Classification and Labelling of Chemicals (GHS), 3rd Revised Edition. United Nations Economic Commission for Europe, Geneva, Switzerland. Accessed 12 November 2013 at http://www.unece.org

UNEP (1987). The Montreal Protocol on Substances that Deplete the Ozone Layer. United Nations Environment Programme, Ozone Secretariat, Nairobi, Kenya. Accessed 12 November 2013 at http://ozone.unep.org.

UNEP (2001). The Stockholm Convention on Persistent Organic Pollutants. United Nations Environment Programme, Secretariat of the Stockholm Convention, Châtelaine, Switzerland. Accessed 12 November 2013 at http://www.pops.int.

UNEP (2013). Report of the Conference of the Parties to the Rotterdam Convention on the Prior Informed Consent Procedure for Certain Hazardous Chemicals and Pesticides in International Trade on the work of its sixth meeting. Conference of the Parties to the Rotterdam Convention on the Prior Informed Consent Procedure for Certain Hazardous Chemicals and Pesticides in International Trade Sixth Meeting. Geneva. Switzerland. Accessed 25 March 2015 at http://www.pic.int.

UNEP (2014). Status of Ratifications. United Nations Environment Programme, Châtelaine, Switzerland. Accessed 3 March 2014 at http://synergies.pic.int.

US EPA (2008). Estimations Programs Interface (EPI) Suite™ for Microsoft Windows®, v 4.10. United States Environmental Protection Agency, Washington DC, USA. Available at http://www.epa.gov.

US EPA (2009). Long-Chain Perfluorinated Chemicals (PFCs) Action Plan. United States Environmental Protection Agency, Washington DC, USA. Accessed 10 December 2014 at http://www.epa.gov.

US EPA (2011). Code of Federal Regulations: Title 40, Section 721.9582 - Certain perfluoroalkyl sulfonates. Accessed 26 February 2015 at http://www.gpo.gov.

US EPA (2014). TSCA Chemical Substance Inventory. United States Environmental Protection Agency, Washington DC, USA. Accessed 26 February 2015 at http://www.epa.gov.

Zou X and Barton SW (1994). X-ray reflection of adsorption of a fluorosurfactant from an organic solution. Langmuir, 10(9),pp 2866-2868.

Share this page